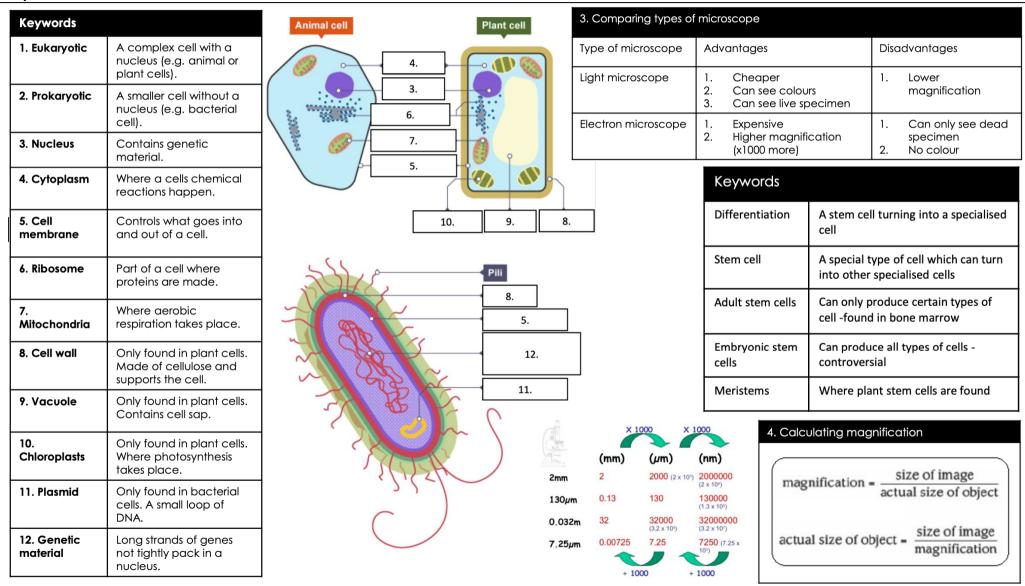
Your teacher will tell you which topic you should revise. Read and learn all the information in the topic, ready for a Quiz in lesson.

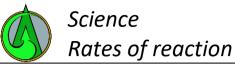
Topic 1:



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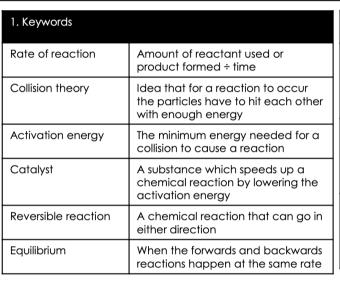
Topic 2:

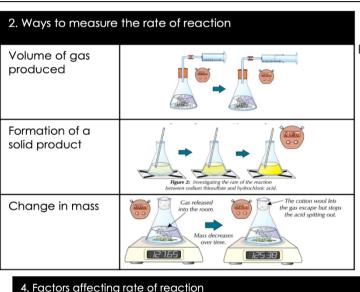
| opic 2: | | | | | |
|---------------------------------------|---|---|---|-------------|---------------------------------------|
| 8. Transport | | | F 1 | Active | ENERGY |
| Keywords Diffusion | | ent of a substance from acentration to an area | Oxygen and carbon dioxide in the lungs Perfume in a room | Transport | High Concentration Low Concentration |
| Osmosis | The movement of water molecules across a partially permeable membrane from a less concentrated solution to a more concentrated solution. | | Water uptake in plants Water absorption in the intestine | Diffusion | High Concentration Low Concentration |
| Active transport | | tance from a lower nigher concentration, ration gradient. Uses | Mineral absorption by roots Glucose absorption by the intestine | Osmosis | |
| Surface area to volume ratio | The surface area div expressed as a ratio | | All high | Osmosis | High Concentration Low Concentration |
| 9. Factors the | at effect the rate of diffusi | ion/osmosis | | - | |
| Speed up | | Slow down | | | |
| High concentration gradient | | Low concentration gradie | ent — | S)→ | Two diploid cells |
| High temperature | | Low temperature | replication | on | |
| High surface area of membrane Low | | Low surface area of mem | nbrane | | Mitosis |

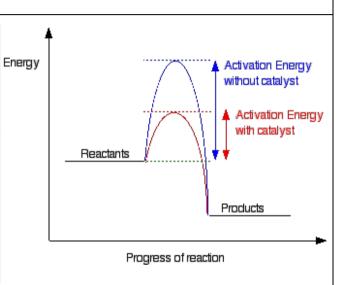


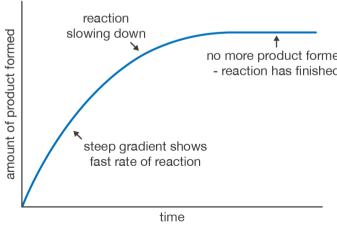
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Topic 3:

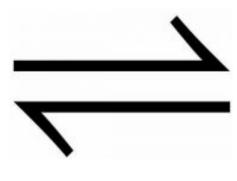


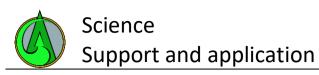






| | | 3 | | |
|-----------|---------------|----------|----------------|--|
| | Factor | Change | Effect on rate | Reason |
| ned ed | Temperature | Increase | Increase | The particles are moving faster so collide more often and with a greater proportion of successful collisions |
| | Concentration | Increase | Increase | The are more particles so collisions are more frequent |
| | Surface area | Increase | Increase | There are more particles available so more collisions |
| | Catalyst | add | increase | The lower activation energy means more particles can successfully collide |





| Vocabulary | Wider Research | Apply |
|--|---|--|
| 1. Eukaryotic | | State one advantage and one disadvantage of using: |
| 2. Prokaryotic | Cells in animals and plants/Cell division & transport | a. A light microscope (2 marks) |
| 3. Nucleus | in cells: | b. An electron microscope (2 marks) |
| Cytoplasm | | |
| Cell Membrane | https://www.bbc.co.uk/bitesize/guides/zwnp7p3/revi | 2. A cell's image has a diameter of 800μm. The actual cell diameter is |
| 6. Ribosome | sion/1 | 20μm. At what magnification is the cell observed? (3 marks) |
| 7. Mitochondria | | |
| 8. Cell Wall | https://www.youtube.com/watch?v=HBZcpzr5B2g&lis | 3. Suggest why the nucleus and the mitochondria are so important in |
| 9. Vacuole | t=PL9IouNCPbCxVU74eQtCcqbaQdYmwzAnlC | cells. (4 marks) |
| 10. Chloroplasts | | |
| 11. Stem Cell | | 4. Describe how active transport differs from diffusion and osmosis (3 |
| 12. Differentiation | Rates of reaction: | marks) |
| 13. Resolution | | |
| 14. Diffusion | https://www.bbc.co.uk/bitesize/guides/zp6xdxs/revisi | 5. Explain why cell division by mitosis is so important in the body (2 |
| 15. Osmosis | <u>on/1</u> | marks) |
| 16. Active Transport | | |
| 17. Mitosis | https://www.youtube.com/watch?v=UkrBJ6- | 6. Explain the difference between embryonic stem cells and adult |
| 18. Meiosis | uGFA&list=PL9IouNCPbCxW8AN0t0py7LaKdKSwfL3fP | stem cells (2 marks) |
| Collision Theory | | |
| 20. Activation Energy | | 7. State how many chromosomes are in a normal human body cell, |
| 21. Catalyst | | and also a human sperm cell (2 marks) |
| 22. Reversible Reaction | | 8. List the factors that can affect the rate of a chemical reaction (4 |
| 23. Equilibrium | | marks) |
| 24. Reactants | | |
| 25. Products | | 9. Explain why the idea of activation energy is an important part of |
| | | collision theory (2 marks) |
| | | |
| | | 10. Explain why food takes longer to cook in a pan than it does a |
| | | pressure cooker (2 marks) |
| | | |
| | | 11. What do chemists mean by the term reversible reaction? (1 mark) |